

# Urine Osmolality and Electrolytes

Why do we do it?

What does it mean?

and What can go wrong?

BCSLS Telehealth Presentation

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# Outline

- Definitions
- Osmometry
- Applications
  - Polyuria
  - Hyponatraemia
  - Renal Failure

# Definition

- Concentration
  - Number of moles per L of solution
- Osmolarity
  - number of osmoles per L of solution
- Osmolality
  - number of osmoles per kg solvent
- Tonicity
  - non-penetrating osmoles
- Plasma osmolarity is 1-2% less than osmolality

# Colligative properties

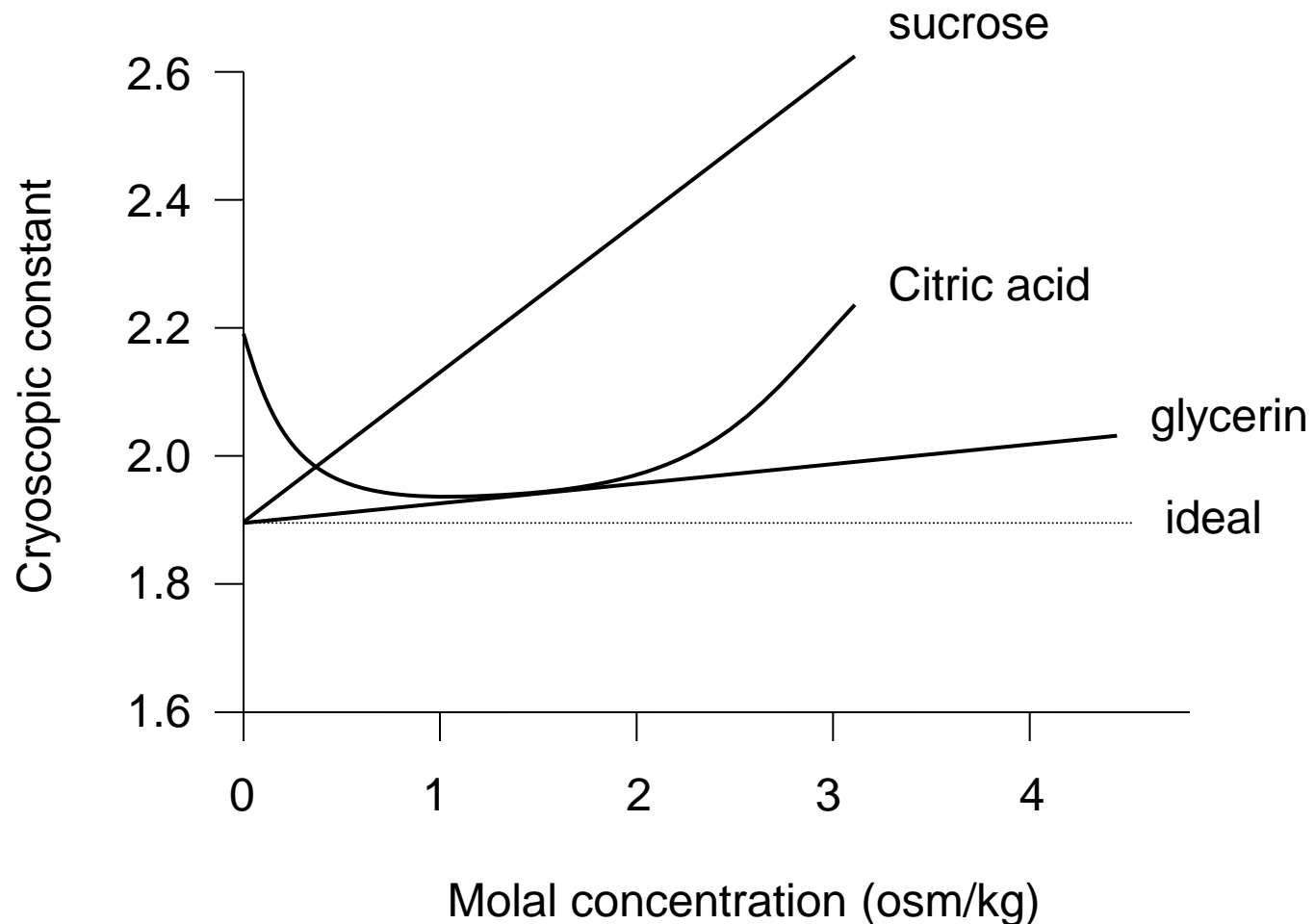
- Parameters proportional to # osmoles:
  - Depressing freezing point
  - Elevating boiling point
  - Increasing osmotic pressure
  - Lowering vapour pressure



# Osmolality Measurement

FP↓	$\Delta T \approx -1.86 \text{ C}^\circ/\text{osm}$	Simple, common
BP↑	$\Delta T \approx +0.52 \text{ C}^\circ/\text{osm}$	Organic compounds unstable.
VP↓	$\Delta P \approx -0.3 \text{ mmHg/osm}$	Volatile chemicals invalidate
OP↑	$\Delta \Pi \approx 17000 \text{ mmHg/osm}$	Technically hard

# Why “ $\approx$ ” in those equations?





Osmometer Ready  
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The Advanced™ Micro Osmometer  
Model 3300

Chamber Cleaners

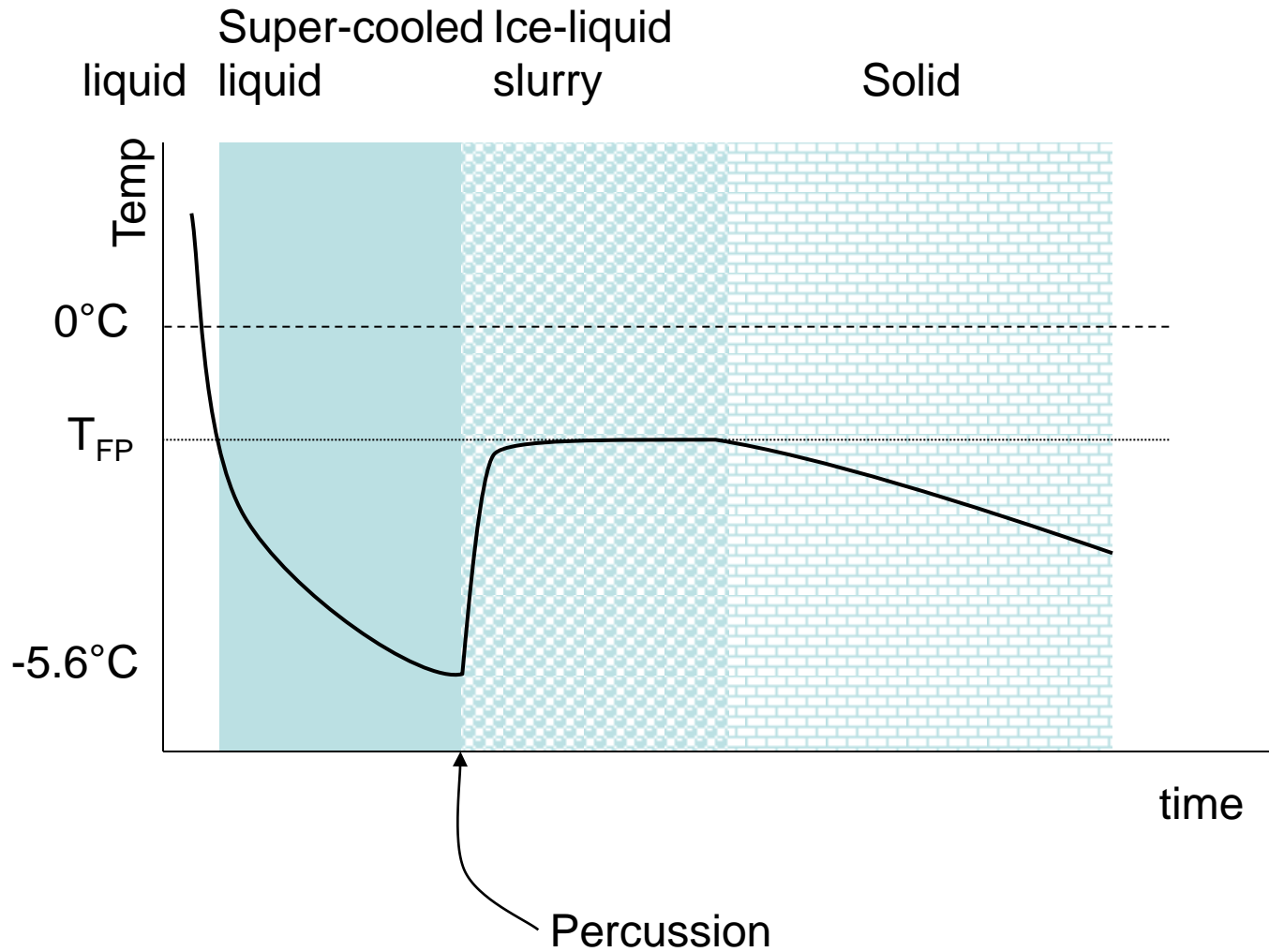
To ensure optimal test results, Advanced Instruments strongly recommends replacing the sampler plunger with (part # 330020) on your 200L sampler unit every 100 samples or with each box of sampler tips.

Sampler Tips

ADVAN INSTR

Health  
ISOPROPR  
RUBBING

# Freezing point osmometry



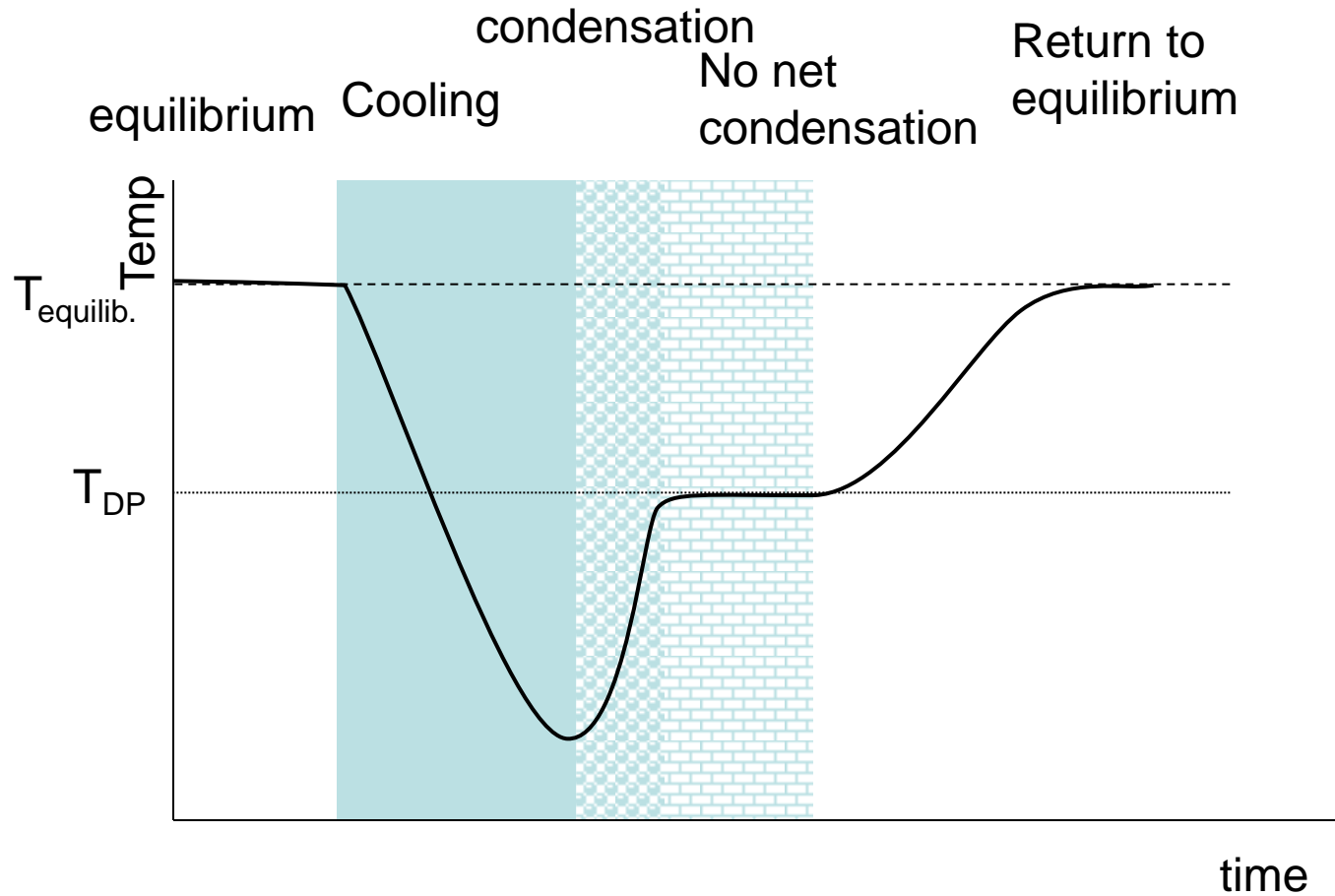
# Freezing Point Analytical Interferents

- High viscosity
- Particles

# Specimen



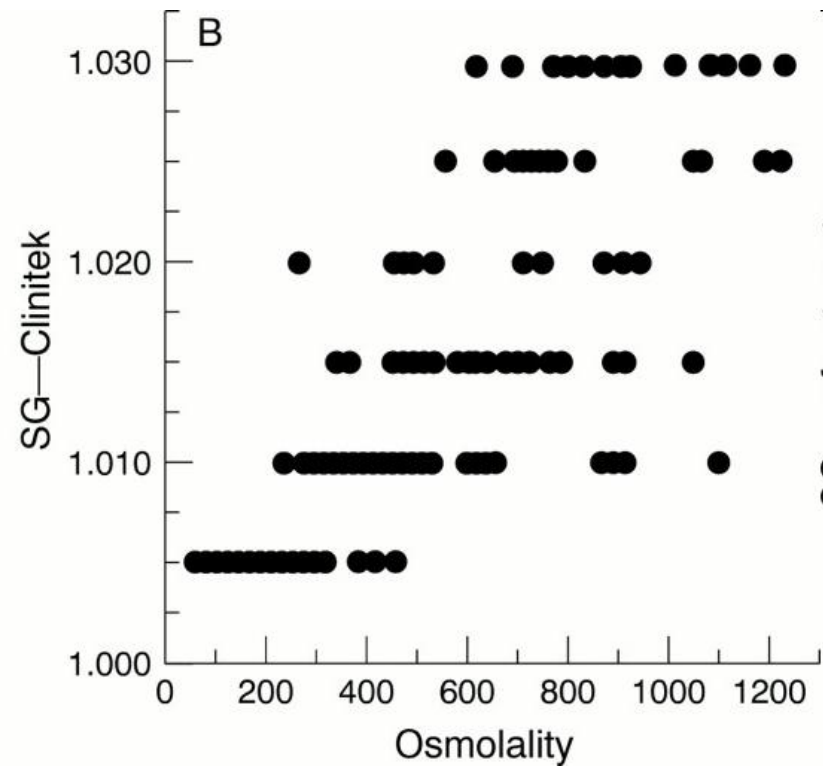
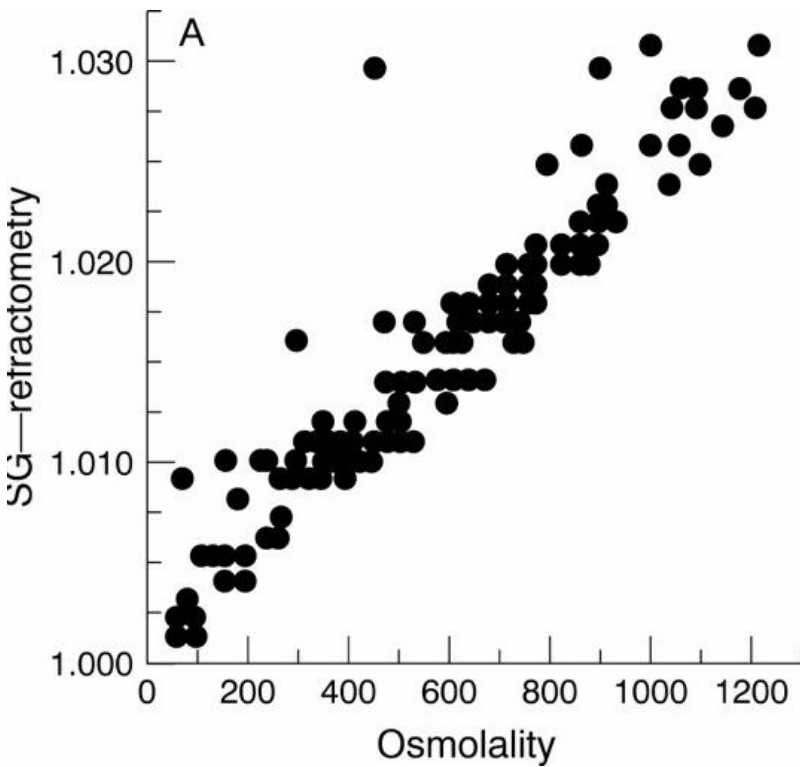
# Dew Point



# Methods

- Freezing point depression
  - Almost exclusive method now
  - Large coefficient so good CV's
  - Simple, robust instruments
- Dew Point
  - Invalid for volatiles

# Alternative Test



→ Refractometry correlates with osmolality

→ weaker correlation with dipstick SG and osmolality

Limitation: glucose and protein lessen relationship

# Specimen

- No special preparation
  - Usually need serum/plasma results
- Random urine
  - Acute conditions/decisions
  - Simple to collect
  - Creatinine improves interpretation
- Timed urine
  - More reliable (Na excretion is variable)

# Normal values

- No normal values for urine electrolytes & osmolality
- Range of physiologically expected results
  - U.osm 50-1200 mosm/kg
  - Range narrows as one ages
- “Expected” values depend on clinical picture

# Urine Osmolality

- Typical values
  - 500-800 mosm/kg (24 hour)
  - 300-900 mosm/kg (random)
  - After 12 hours fasting  $>850$  mosm/kg
  - A.m. specimen  $\approx 3$  times serum osmolality

# Urine Osmolality

- Increased
  - Dehydration
  - SIADH
  - Adrenal insufficiency
  - Glycosuria
  - Hypernatraemia
  - High protein diet
- Decreased
  - Diabetes insipidus
  - Excess fluid intake
  - Renal insufficiency
  - glomerulonephritis

# Urine Osmolality Cautions

- EtOH
- Extremes of protein in diet
- Diuretics

# Other Specimens

- Pleural, ascitic
  - In equilibrium with serum
  - Thus, equal to serum
  - Rarely indicated
  - No special requirements for measurement

# Faecal specimens

- Rare test for cause of diarrhoea
- $F_{osm} - 2(F_{Na} + F_{K})$ 
  - $<50$ : non-osmotic
  - $>150$ : osmotic diarrhoea
    - Laxatives
    - Malabsorption (including lactose intolerance)
    - Sorbitol, mannitol, lactulose
- Limitation:
  - bacterial activity ↑'s gap
  - Only very watery specimens

# Clinical Indications

- Polyuria
- Hyponatraemia
- Renal failure
- Low anion-gap metabolic acidosis

Polyuria

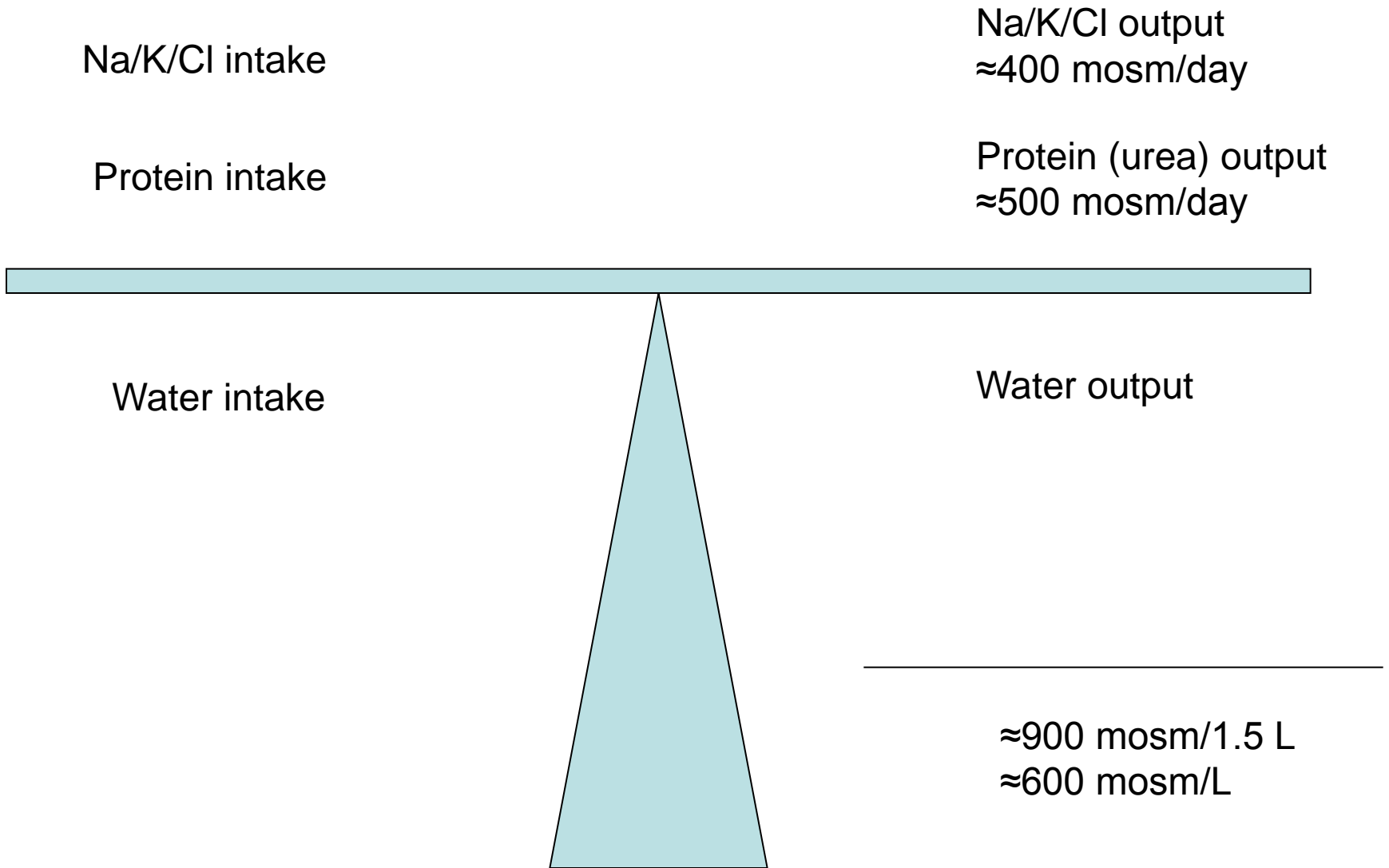
# Case

- 65 year old
- c/o going to the w/c 4 or 5 times/night
- Constantly thirsty
- Drinks 4 or 5 L of ice cold water/day
- Urine output measured at 3.5 L/day

# Normal Urine Output

- $<50$  mL/day                      anuria
- $<500$  mL/day                      oligouria
- $>3$ L /day                              polyuria
  
- Depends on individual
  - If hypernatramic or hypovolaemic  
     $>800$  mL/d is polyuria

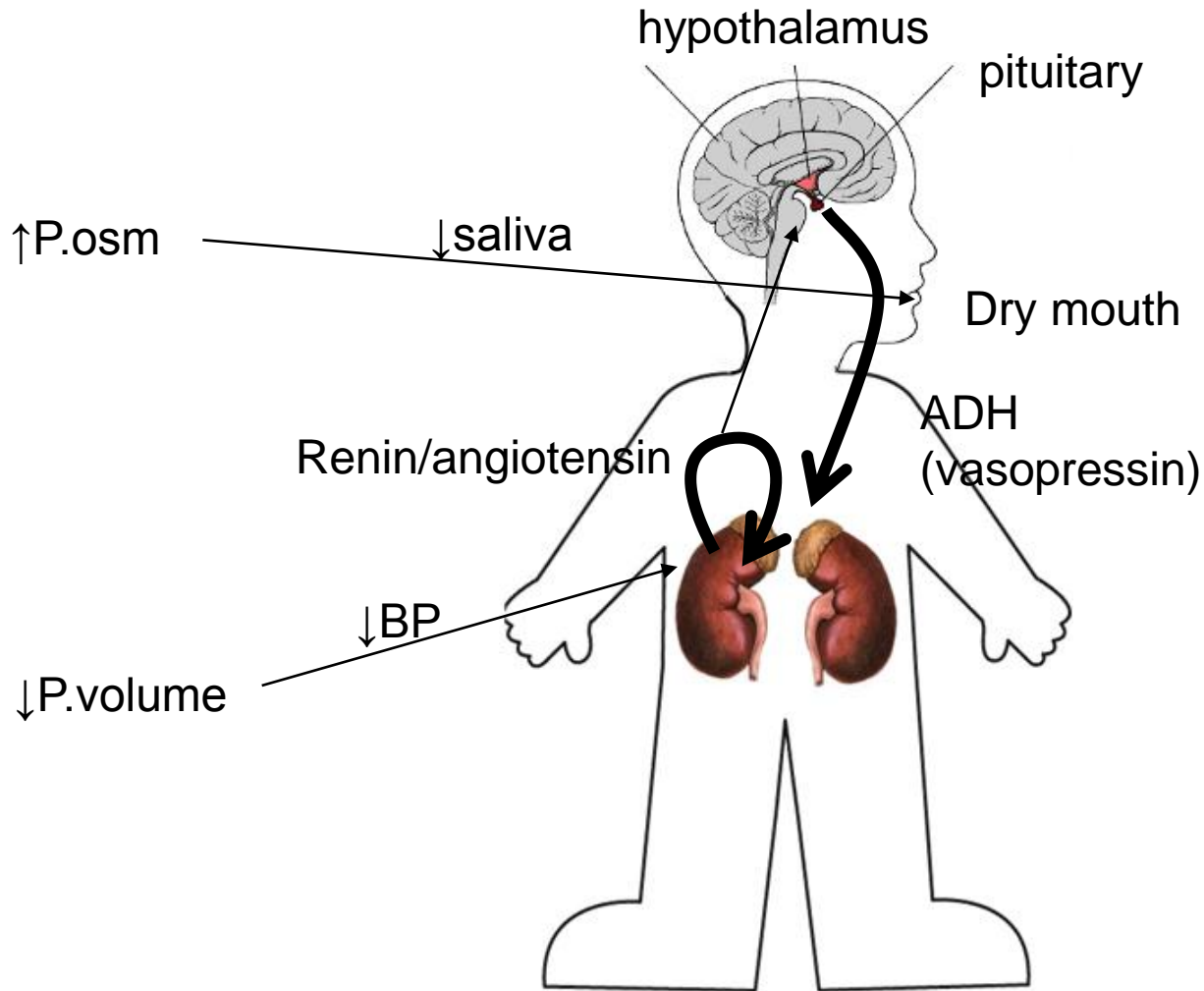
# Water & Electrolyte Balance



# Water Balance

- Metabolism 10%  
250mL
- Food 30% 750mL
- Beverages 60% 1500 mL
- Faeces 4% 100 mL
- Sweat 8% 200 mL
- “Insensible” (skin + lungs) 28% 700 mL
- Urine 60% 1500 mL

# Water Regulation

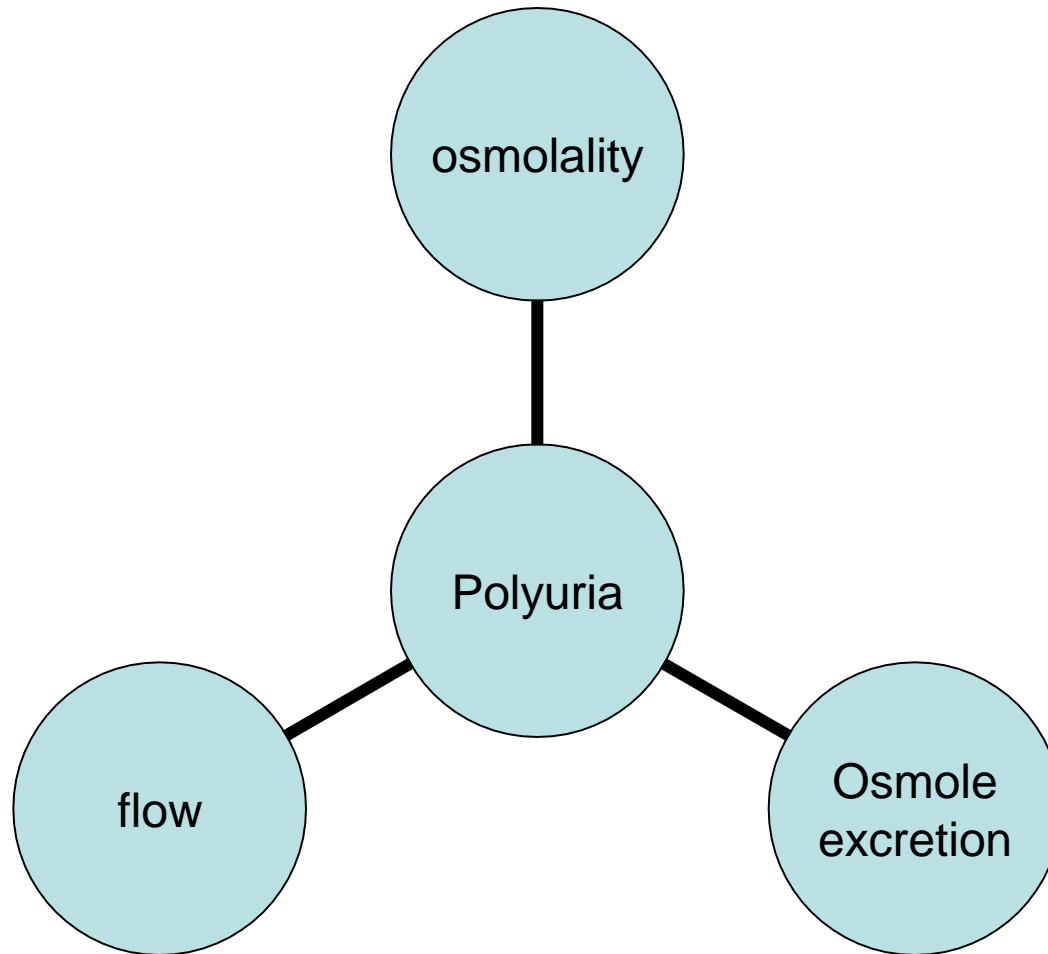


# Assessing Intravascular Volume

- Clinical (sensitivity ~50%)
  - Orthostatic hypotension
  - Weak pulse
  - Cool extremities
  - HR
  - Skin turgor
  - Mucous membranes
  - JVP
- Investigation
  - Urine Volume & Colour
  - Chest X-ray
  - Haemoconcentration
  - Creatinine, urea
  - Radioactive albumin dilution
  - BNP



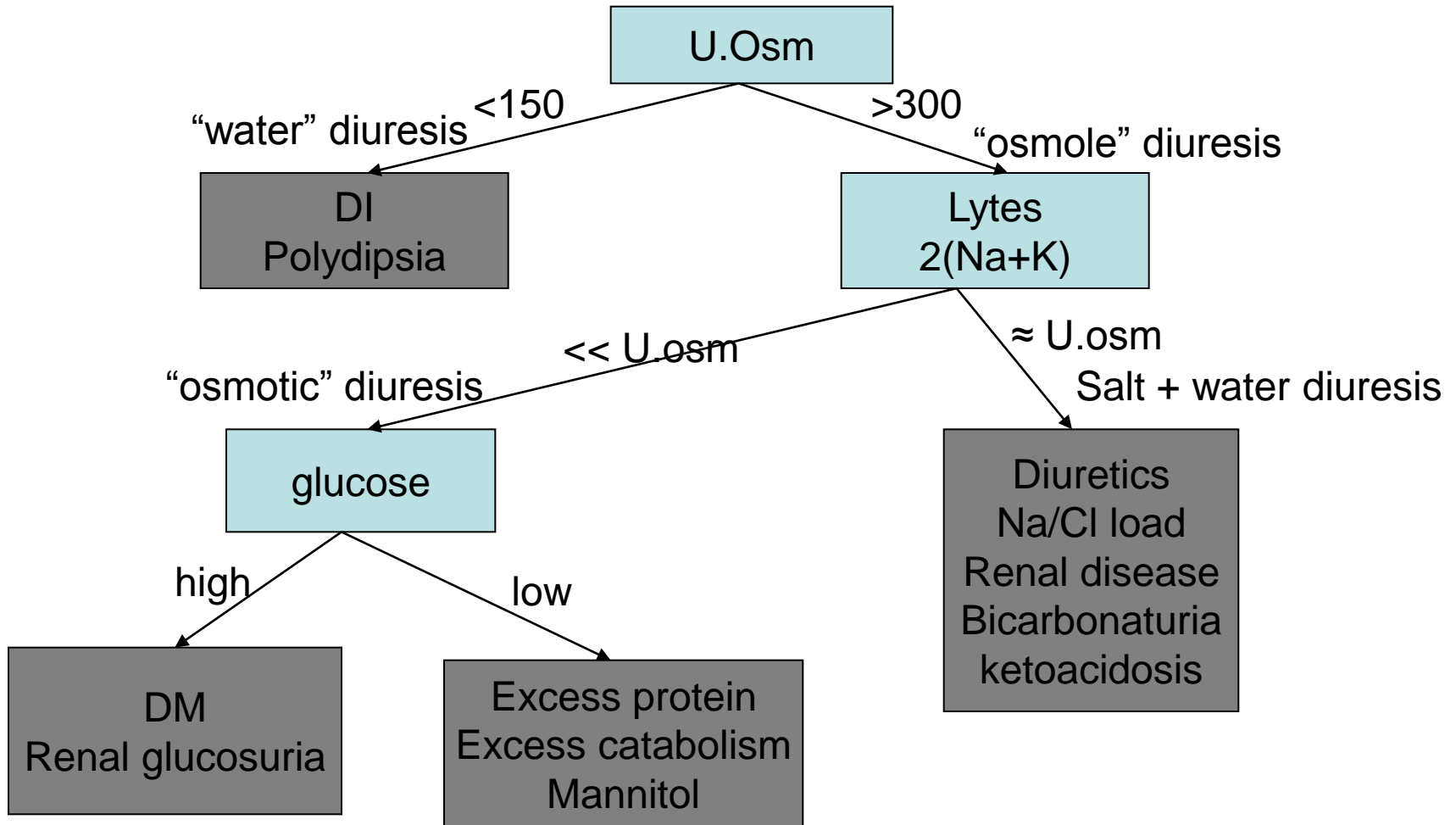
# Polyuria Assessment



# Polyuria

- $<150$  mosm/kg
  - Water diuresis
    - Diabetes Insipidus (P.Na high/normal)
    - Polydipsia (P.Na low/normal)
- $>300$  mosm/kg
  - Solute diuresis

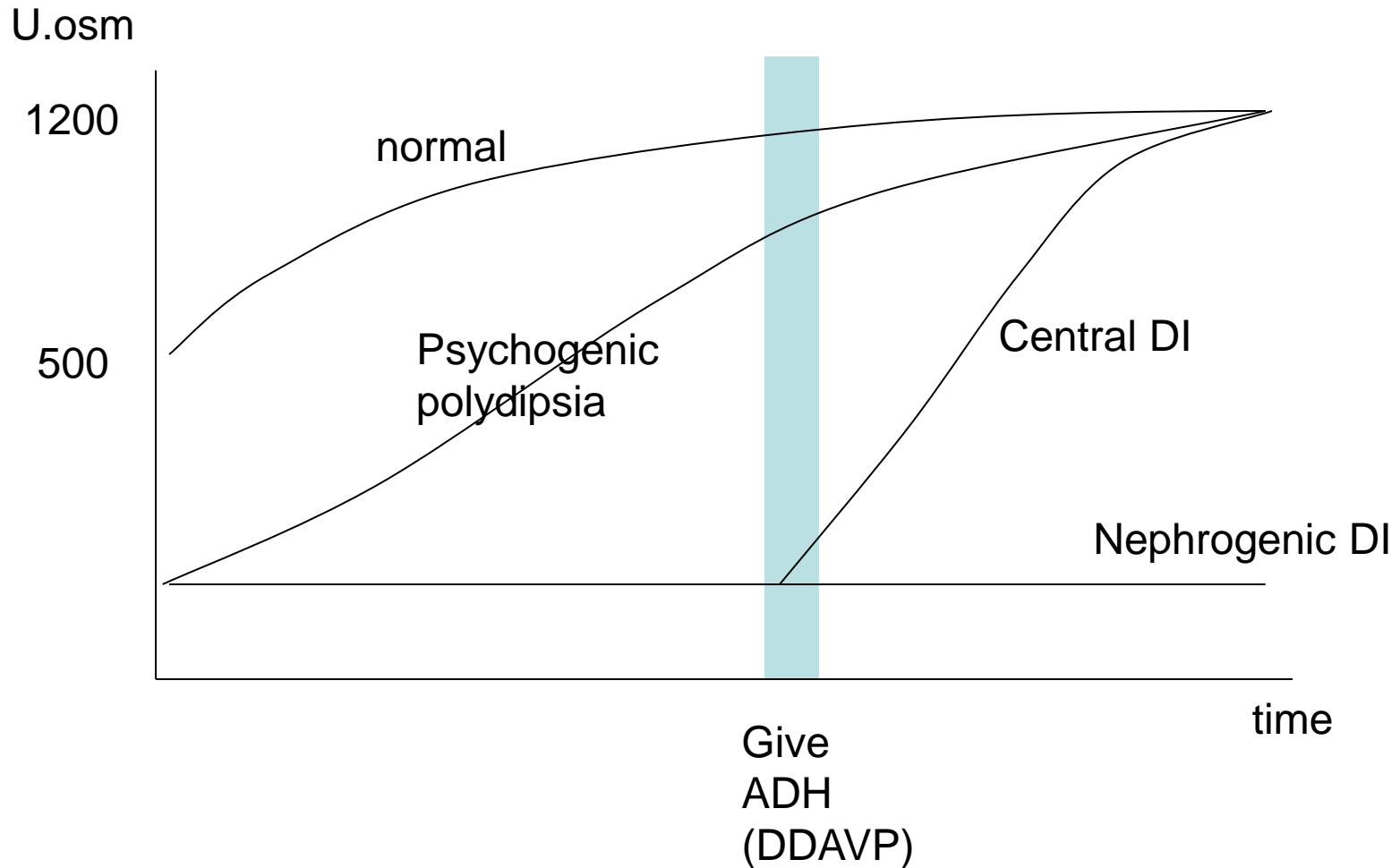
# Polyuria



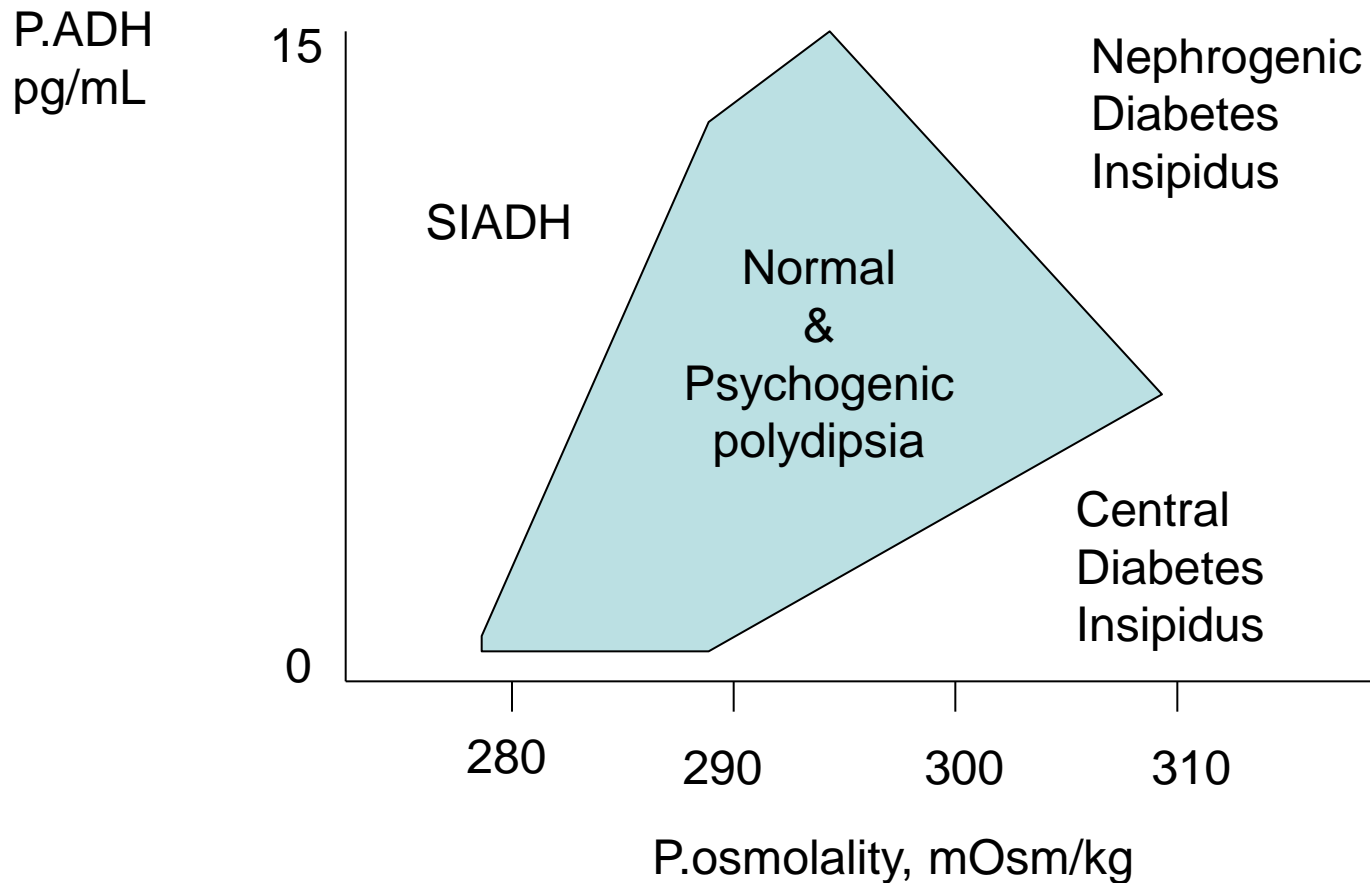
# Water Deprivation Test

- Preparation: Patient drinks until 6:00 am morning of test
- Weight, P.osm, P.Na, U.osm, U.Na
- Nothing to eat/drink during test
- Urine hourly
- Stop test if U.osm  $> 500$ , P.Na  $> 145$ , weight loss  $> 10\%$ .
- +/- DDAVP

# Water Deprivation Test



# ADH & plasma osmolality



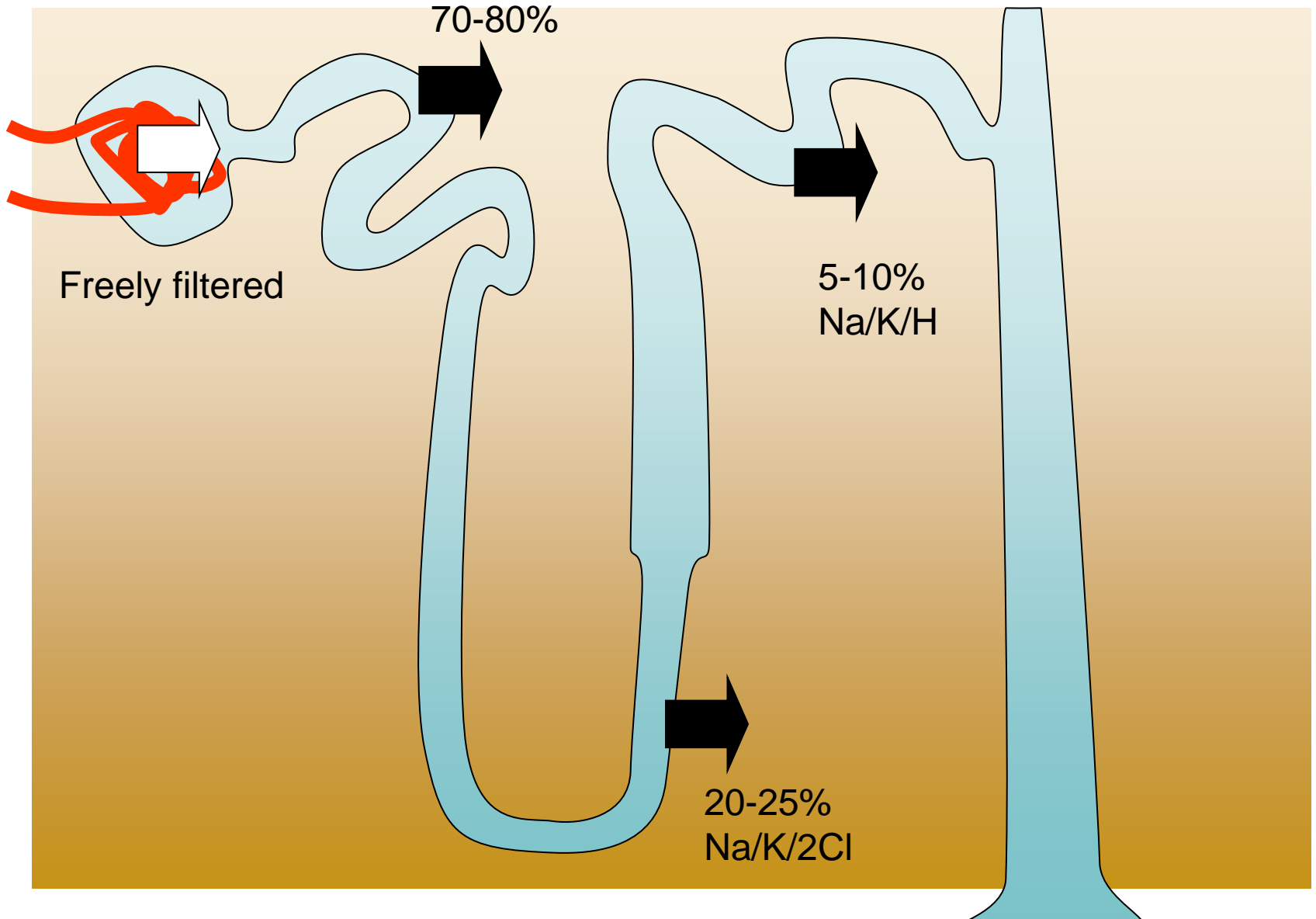
# Hyponatraemia

# Renal Handling

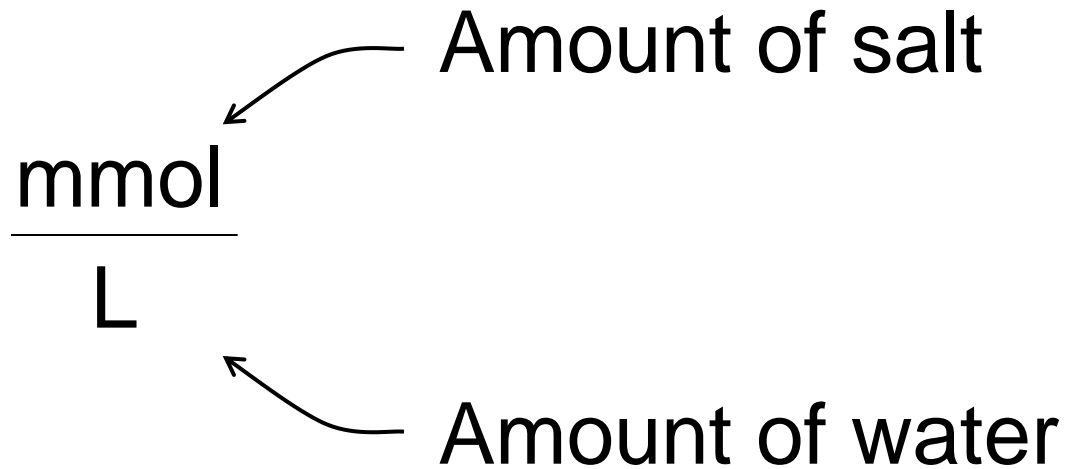


- Average 70 kg male filters approximately 1.2 kg of salt per day!
- Vast majority must be reabsorbed

# Urine Sodium



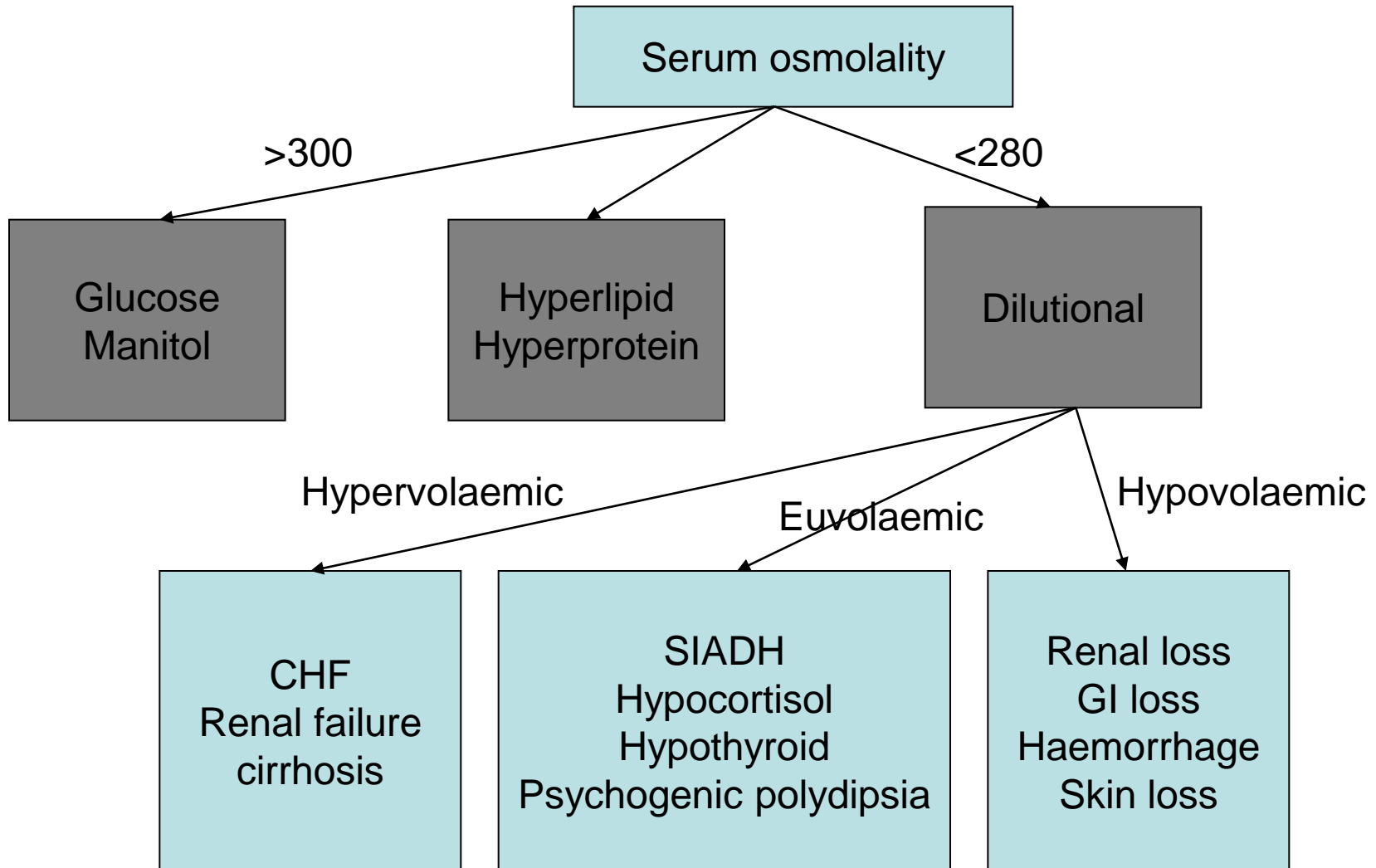
# Dysnatraemia



# Case

- 6 month old with astrocytoma on vincristine
- P.Na 126 L 135-145 mmol/L
- P.Osm 255 L 280-300 mosm/kg  
U.Na 32
- Cerebral salt wasting vs SIADH

# Hyponatraemia



# Case...

- SIADH:
  - Kidneys act to preserve perfusion
  - SIADH = excess of water, hence intravascular volume, in face of low Na.
  - Kidney response is to lose Na
- Cerebral Salt Wasting
  - Inappropriate Na excretion →
    - ↑urine volume →
    - ↓intravascular volume

# SIADH

- Laboratory Features
  - ↓S.Na
  - ↓P.osm
  - ↑U.osm (typ. >50 mosm/L)
  - U.Na >20 mmol/L
  - Normal renal, thyroid & adrenals
  - Euvolaemic

# SIADH Causes

- Tumour
  - Small cell, bronchogenic CA, pancreatic, Hodgkin's
- Pulmonary
  - Pneumonia, lung abscess, TB
- Medications
  - NSAID, barbiturates, carbamazepine, TCA, oxytocin
- CNS
  - Brain tumour, encephalitis, SAH, AIP, trauma
- AIDS
- Ventilation
- Post-operative

# SIADH Treatment

- Water restriction (~1 L/day)
- Salt
- Weigh daily
- Loop diuretic
- Urea (rarely used)

# Cautions

- No renal or adrenal problems
- No bicarbonaturia
  - Obligate excretion of Na
  - E.g. recent vomiting
- No carbonic anhydrase inhibitors
- No acid/base disturbance
- No diuretics

# Hyponatraemia

- Reset osmostat
  - Test: water load to further drop S.Na
  - Rarely performed test
  - Only done in patients with mild hyponatraemia
  - SIADH → urine remains concentrated
  - Reset osmostat → urine becomes dilute
- Importance: treatment different

# Treatment cautions

- Dangerous to correct too correctly
- Central pontine myelinolysis
- $\uparrow$ Na by 1-2 mmol/L/h,  
maximum of 8 mmol/L/day

# Hypernatraemia

- Very rarely *need* urine studies
  - Can be used as an adjunct study

# Urine Potassium

- Less predictable due to many influences
  - Potassium intake variable
  - Na effects via aldosterone
  - Hydration
- Range 10-400 mmol/d
- Transtubular potassium gradient
  - Can be calculated
  - Clinically doesn't add much information
  - Rarely used now

# Renal Failure

# Renal Failure Classifications

- Duration
  - Acute renal failure
  - Chronic renal failure
- Location
  - Pre-renal
  - Renal
  - Post-renal

# Renal Failure

- Pre-renal
  - Dehydration
  - CHF
  - Arterial supply
- Renal
  - Glomeruli
  - Renal tubules
  - Interstitium
- Post-renal
  - Stone
  - Tumour
  - Prostate

# Fractional Excretion

- Corrects for urine concentration
- $FE_{Na} = [U.Na * P.Cr] / [U.Cr * P.Na]$
- N.B. match units
  - Plasma creatinine in  $\mu\text{mol/L}$
  - Urine creatinine in  $\text{mmol/L}$
- $FeNa < 1\%$     pre-renal
- $FeNa > 1\%$     renal

# Case #1

- 50 y.o. goes to the ER with history of vomiting and diarrhoea. Looks dehydrated

Plasma.Na    140                    135-145 mmol/L

Plasma.Cre    150 HΔ                    umol/L

U.Na                    25                    mmol/L

U.Cre                    1                    mmol/L

FENa                    2.7%

U.osm                    320                    mosm/kg

# Case #2

- 50 y.o. goes to the ER with history of vomiting and diarrhoea. Looks dehydrated

Plasma.Na    140                    135-145 mmol/L

Plasma.Cre    150  $\mu$ mol/L

U.Na                    25                    mmol/L

U.Cre                    10                    mmol/L

FENa                    0.27%

U.osm                    600                    mosm/kg

# Interpretation

- FENa  $<1\%$  -- pre-renal
- FENa  $>3\%$  -- acute kidney injury
- U.Osm  $>500$  – pre-renal
- U.Osm  $<350$  – acute kidney injury
  
- No gold-standard test
- Acute kidney injury can develop from pre-renal failure

# Caution

- Loop diuretics increase FENa
  - Cannot interpret while on diuretics
- Severe protein malnutrition/starvation or wash-out of kidney

RTA

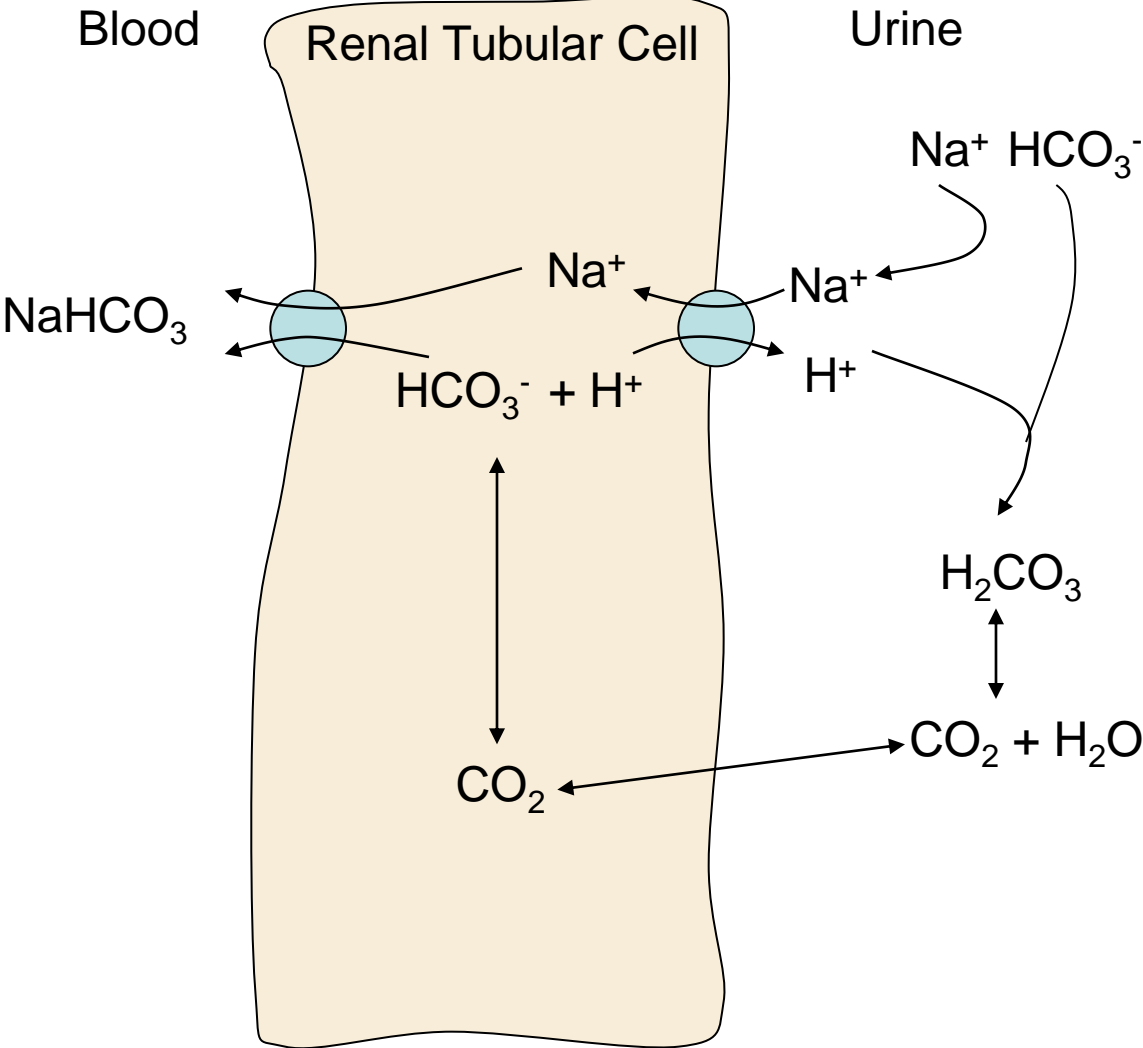
# Metabolic Acidosis

- Increased anion gap
  - Relatively straight-forward
  - “MUDPILES”
- Normal anion gap
  - Factitious
    - Hypoalbuminaemia
    - Unusual cations (e.g. monoclonal band)
  - HCl or NH<sub>4</sub>Cl loading
  - HCO<sub>3</sub> loss
    - GI (diarrhoea, ileus)
    - Renal (proximal RTA, carbonic anhydrase inhibitor)
  - Failure to generate “new” bicarbonate (distal RTA)
  - Gain of an acid with excretion of conjugate base

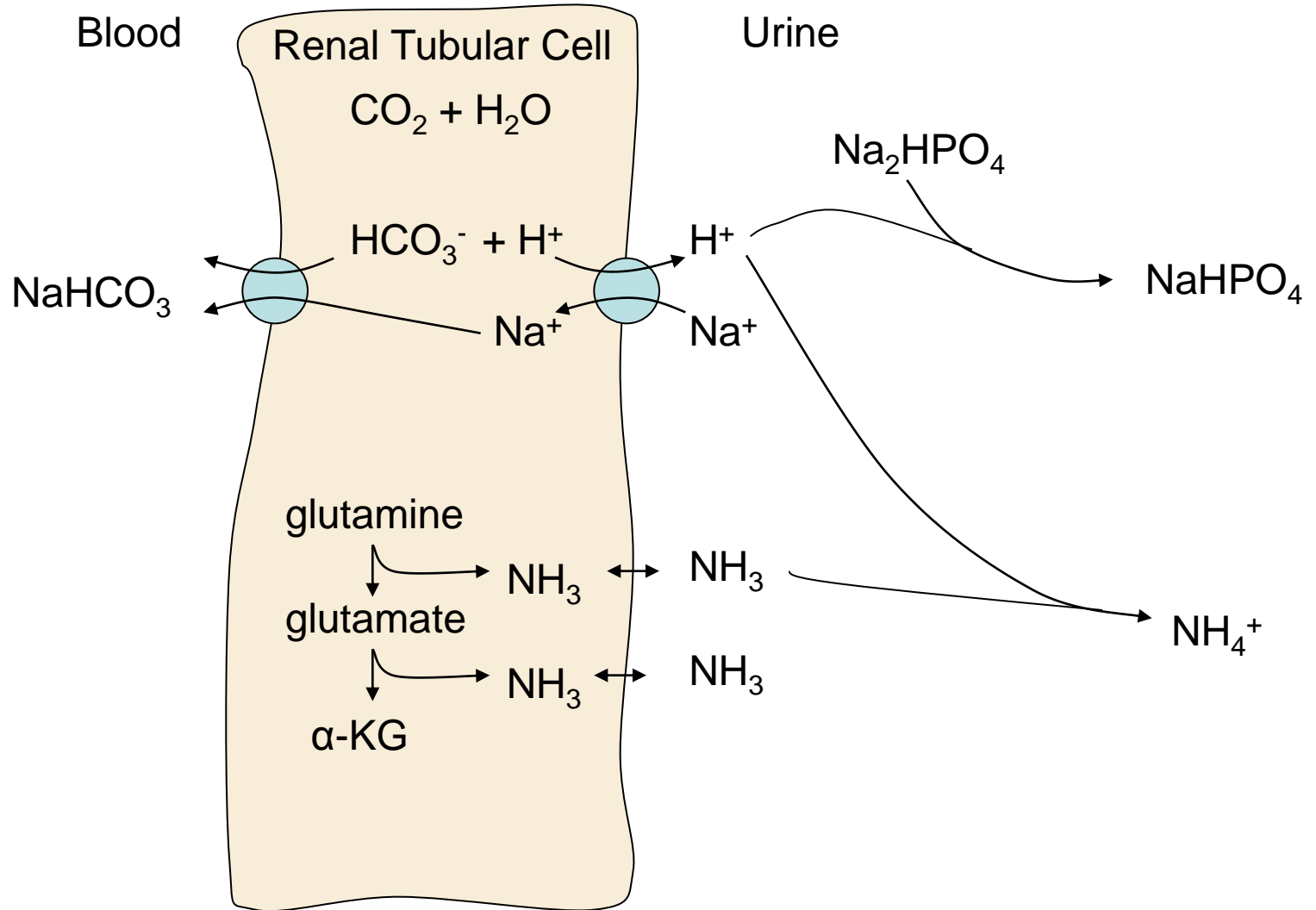
# Acid production

- Typical North American diet produces 1 mmol H<sup>+</sup>/kg body weight/day
- Mainly from protein oxidation
- Buffering acid load consumes equivalent amount of bicarbonate.
- S.pH ~7.4, U.pH ~6.0

# Renal Acid-Base: Recovery



# Renal Acid-Base: Loss of H<sup>+</sup>



# Ammonia

- Theoretically can be measured; however, technically very difficult in urine
- Essentially research-only method
- Calculating

# Electroneutrality

+ Na<sup>+</sup>

+ K<sup>+</sup>

+ 2Ca<sup>++</sup>

+ 2Mg<sup>++</sup>

+ NH<sub>4</sub><sup>+</sup>

+ Cl<sup>-</sup>

+ HCO<sub>3</sub><sup>-</sup>

+ H<sub>2</sub>PO<sub>4</sub><sup>-</sup>

+ 2HPO<sub>4</sub><sup>=</sup>

+ 2SO<sub>4</sub><sup>=</sup>

+ organic anions

# Electroneutrality



≈80 mEq/day



+ organic anions

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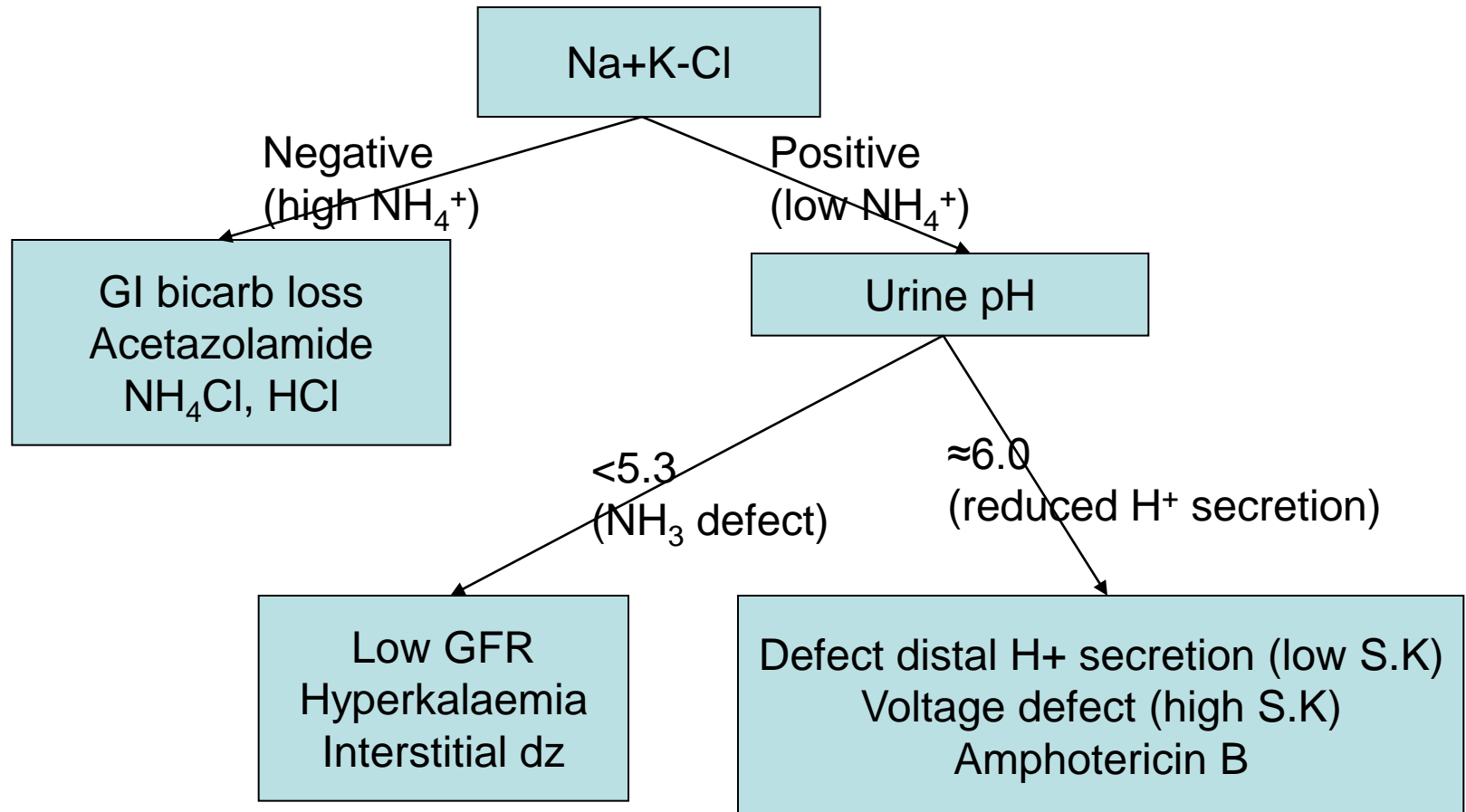
$$\text{Na}^+ + \text{K}^+ + \text{NH}_4^+ = \text{Cl}^- + 80$$

$$\text{NH}_4^+ \propto \text{Na}^+ + \text{K}^+ - \text{Cl}^-$$

# Cautions of net charge

- Unmeasured anions → underestimate  $\text{NH}_4$ 
  - Ketonuria
  - DKA
  - Drugs (penicillin, salicylates)

# Non-Anion gap Metabolic Acidosis



# Conclusions

# Urine Osmolality & Lytes Utility

- Rarely needed, but critical test
  - Polyuria
  - Hyponatremia
  - Supportive role
    - Pre-renal vs renal oliguria
    - Integrity of the medullary interstitium

# Summary

- No “normal” ranges – interpret in clinical context
- Cautions:
  - No diuretics
  - No adrenal or thyroid disease
  - Relatively normal diet